## Short Communication

Laser-induced time-resolved phosphorescence of 2,4-pentadione (a tautomeric system)\*

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The quenching reactions of electronically excited  $\alpha$ -diketones have been reported frequently in the literature. In particular biacetyl has been studied by numerous workers because of its utility in photosensitized emission studies. However, there has not been much attention given to the quenching of excited  $\beta$ -diketones. Tautomerism can occur easily with  $\beta$ -diketones and thus the self-quenching reaction of the triplet state of these molecules may be enhanced by the participation of the unsaturated enolic form as a quencher. For biacetyl it should be noted that at 25 °C there is no observable selfquenching of the  ${}^{3}A_{u}$  state [1 - 4]. Thus, we have undertaken the present investigation to see whether 2,4-pentadione (PD) exhibits self-quenching and to compare its dynamical properties with other diketones.

## 1. Experimental

A Phase-R Model 2100B coaxial flashlamp-pumped dye laser was operated with Pilot "386" dye in a flat-flat cavity. The lasing wavelength was 3891 Å with an FWHM of 40 Å. The laser beam was passed through an appropriate arrangement of spatial filters and neutral density filters before entering the 20 cm diameter and 30 cm length cylindrical quartz photolysis cell. Emission was viewed at right angles to the cell with an RCA 7265 photomultiplier tube. A filter transmitting at  $\lambda$  greater than 4500 Å was placed in front of the photomultiplier tube to discriminate against scattered laser light. Neutral density filters were used to reduce the intense laser beam to ensure that the concentration of triplet PD molecules was low enough to avoid triplet-triplet annihilation reactions such as those noted for intense laser irradiation of biacetyl [1]. All decays noted were single exponentials. The output from the photomultiplier tube was properly terminated at the input of a Tektronix Model 556 oscilloscope and the displayed decay trace was photographed using high speed Polaroid film. A mercury-free grease-free

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vacuum system fitted with the appropriate pressure gauges was used to handle the gaseous components. The PD was the high purity product of the Eastman Kodak Company and the  $O_2$  and Ar were the high purity products of Matheson and Airco, respectively. The PD was bulb-to-bulb distilled and the middle fraction was retained for experiments. The other gases were used without further purification.

## 2. Results and discussion

Figure 1 displays a Stern-Volmer plot for PD self-quenching. Figure 2 depicts the Stern-Volmer plots for  $O_2$  and Ar quenching. Note in Fig. 1 that PD does exhibit self-quenching and the quenching rate constant is estimated to be  $2.2 \times 10^5 \ \text{l} \ \text{mol}^{-1} \ \text{s}^{-1}$ . The collision-free lifetime for the emission is 1.8 ms. In Fig. 2 it is noted that Ar does not have a noticeable quenching effect at all on the emission from electronically excited 2,4-pentadione (\*PD). However, \*PD is quenched by  $O_2$  relatively efficiently and its quenching constant is estimated to be  $6.6 \times 10^8 \ \text{l} \ \text{mol}^{-1} \ \text{s}^{-1}$ . The relative quenching rates by Ar and  $O_2$  would support the emission as being phosphorescence.

Biacetyl  ${}^{3}A_{u}$  molecules do not exhibit self-quenching at room temperature as do the  ${}^{3}PD$  molecules. One possibility which would tend to lead to an increased self-quenching efficiency for  ${}^{3}PD$  is that PD displays tautomerism in which 91 - 93% of PD is present in the enolic form in the gas phase at 25 °C [5]. Even though the  ${}^{3}PD$  may conceivably have a lower triplet level than enolic PD, the formation of a charge-transfer complex between the  $\pi$ systems of enolic PD and  ${}^{3}PD$  could still facilitate a reasonable quenching efficiency, as has been noted in other triplet quenching reactions [6, 7].

The magnitude of the quenching constant of <sup>3</sup>PD by O<sub>2</sub>,  $6.6 \times 10^8$  l mol<sup>-1</sup> s<sup>-1</sup>, compares closely with rate constant values of  $5.6 \times 10^8$  l mol<sup>-1</sup> s<sup>-1</sup> [8] and  $5.4 \times 10^8$  l mol<sup>-1</sup> s<sup>-1</sup> [9] reported for quenching of biacetyl <sup>3</sup>A<sub>u</sub> molecules. It may be that the detailed mechanistic quenching reaction

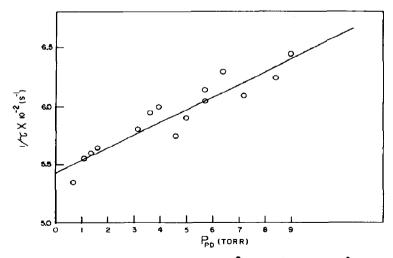


Fig. 1. A Stern–Volmer plot at 25 °C and  $\lambda$  = 3891 Å for PD self-quenching.

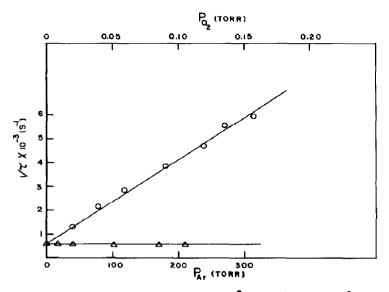


Fig. 2. A Stern-Volmer plot at 25 °C and  $\lambda$  = 3891 Å for quenching of electronically excited PD by O<sub>2</sub> (circles) and Ar (triangles); P<sub>PD</sub> = 5.70 Torr.

by  $O_2$  is quite similar for <sup>3</sup>PD and <sup>3</sup> $A_u$  biacetyl molecules. There are no known chemical quenching routes involving Ar, and the quenching rate by a physical energy-transfer mechanism generally correlates with the energy level separation of the collision partners. The fact that Ar collisions are ineffective in relaxing triplet PD indicates that this triplet energy level is relatively low-lying. Further work is needed in the gas phase to understand properly the excited state dynamics of compounds that can undergo tautomerism.

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